

4-Acetyl-*N,N*-dibenzoylphenylamine

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## Key indicators

Single-crystal X-ray study

$T = 183\text{ K}$

Mean  $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$

$R$  factor = 0.054

$wR$  factor = 0.135

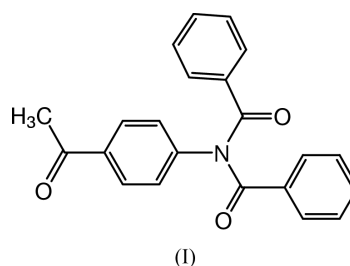
Data-to-parameter ratio = 14.3

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title compound,  $\text{C}_{22}\text{H}_{17}\text{NO}_3$ , the three aromatic rings are orthogonal. The crystal structure is stabilized by weak intermolecular  $\text{C}-\text{H}\cdots\text{O}$  and  $\text{C}-\text{H}\cdots\pi$  interactions.

## Comment

In recent years, photo-induced electron transfer (PET) reactions of organic compounds have been extensively investigated (Scheinbaum, 1964; Kavarnos & Turro, 1986; D'Auria *et al.*, 1996). Such PET reactions of nitroaromatic compounds, which are electron acceptors, with a variety of donors are also of interest. We have recently investigated the photo-induced reactions between nitroaromatic compounds and diphenylacetylenes, which give the corresponding benzenamines as one of the products. The crystal structure of *N,N*-dibenzoyl-4-chloroaniline, which resulted from such a reaction between 4-chloronitrobenzene and an excess amount of 1,2-diphenylacetylene, has been reported (Usman *et al.*, 2002). We carried out a similar reaction for 4-carbomethoxynitrobenzene and isolated the title compound, (I), which was structurally analysed in order to elucidate its conformation.



The bond lengths and angles in (I) have normal values (Allen *et al.*, 1987), and are comparable with those of the related compound (Usman *et al.*, 2002). Also similar is the orthogonal disposition of the three aromatic rings, with dihedral angles between  $\text{C}1-\text{C}6$  and  $\text{C}8-\text{C}13$ ,  $\text{C}1-\text{C}6$  and  $\text{C}15-\text{C}20$ , and  $\text{C}8-\text{C}13$  and  $\text{C}15-\text{C}20$  of  $76.8(1)^\circ$ ,  $78.8(1)^\circ$ , and  $80.7(1)^\circ$ , respectively. The benzoyl moieties  $\text{C}7-\text{C}13/\text{O}1$  and  $\text{C}14-\text{C}20/\text{O}2$  attached at atom  $\text{N}1$  have the planes of their amide groups,  $\text{O}1/\text{C}7/\text{C}8/\text{N}1$  and  $\text{O}2/\text{C}14/\text{C}15/\text{N}1$ , twisted from the aromatic planes by angles of  $33.2(1)^\circ$  and  $25.6(1)^\circ$ , respectively.

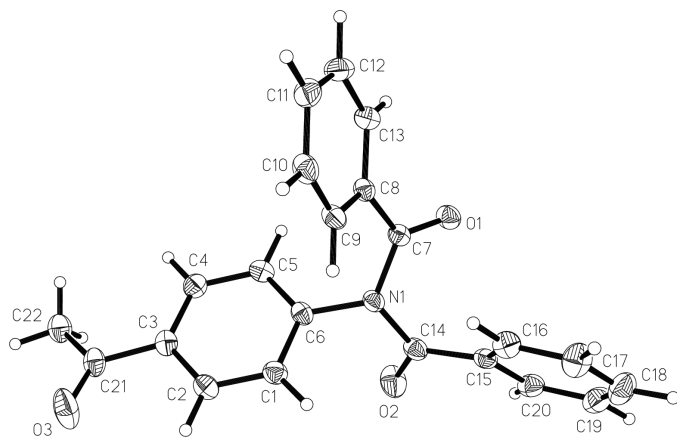
The angles subtended at atom  $\text{N}1$  ( $sp^3$  state) joining the 4-acetylphenyl and benzoyl moieties are in the range  $116.7(1)-121.0(1)^\circ$ . The  $\text{C}21/\text{C}22/\text{O}3$  acetyl group is twisted around the  $\text{C}3-\text{C}21$  bond by  $14.2(1)^\circ$  with respect to the attached aromatic ring.

In the crystal packing, there are two weak  $\text{C}-\text{H}\cdots\pi$  interactions involving the centroids of the aromatic rings

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**Figure 1**

The structure of the title compound, showing 50% probability displacement ellipsoids and the atom-numbering scheme.

C8–C13 [Cg(B)] and C15–C20 [Cg(C)] (Table 2). The molecules are also joined by weak intermolecular C–H...O contacts (Table 2). The C22–H22B...O1<sup>iii</sup> interaction interconnects adjacent molecules into dimers which are linked into a three-dimensional molecular network by C13–H13...O3<sup>ii</sup> and C9–H9...O1<sup>i</sup> interactions (Fig. 2; see Table 2 for symmetry codes).

## Experimental

The title compound was prepared by the photo-induced reaction of an acetonitrile solution of 4-carbomethoxynitrobenzene, with an excess of 1,2-diphenylacetylene. The compound is one of the products of this photo-induced reaction and was isolated by column chromatography on silica gel. Single crystals were grown by slow evaporation from petroleum ether–ethyl acetate.

### Crystal data

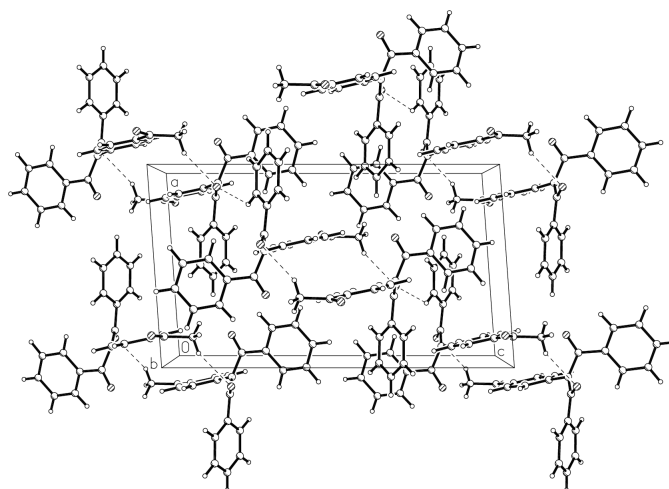
C <sub>22</sub> H <sub>17</sub> NO <sub>3</sub>	$D_x = 1.307 \text{ Mg m}^{-3}$
$M_r = 343.37$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/n$	Cell parameters from 7550 reflections
$a = 10.5190 (1) \text{ \AA}$	$\theta = 3.0\text{--}28.3^\circ$
$b = 9.1443 (2) \text{ \AA}$	$\mu = 0.09 \text{ mm}^{-1}$
$c = 18.1947 (3) \text{ \AA}$	$T = 183 (2) \text{ K}$
$\beta = 94.148 (1)^\circ$	Block, colorless
$V = 1745.54 (5) \text{ \AA}^3$	$0.46 \times 0.30 \times 0.18 \text{ mm}$
$Z = 4$	

### Data collection

Siemens SMART CCD area-detector diffractometer	4192 independent reflections
$\omega$ scans	3124 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$R_{\text{int}} = 0.074$
$T_{\text{min}} = 0.961$ , $T_{\text{max}} = 0.985$	$\theta_{\text{max}} = 28.3^\circ$
10245 measured reflections	$h = -11 \rightarrow 13$
	$k = -12 \rightarrow 11$
	$l = -22 \rightarrow 24$

### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0621P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.054$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.135$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 0.96$	$\Delta\rho_{\text{max}} = 0.49 \text{ e \AA}^{-3}$
4192 reflections	$\Delta\rho_{\text{min}} = -0.37 \text{ e \AA}^{-3}$
293 parameters	Extinction correction: <i>SHELXTL</i>
H atoms treated by a mixture of independent and constrained refinement	Extinction coefficient: $0.037 (3)$



**Figure 2**

Packing diagram of the title compound, viewed down the  $b$  axis, showing the intermolecular contacts linking molecules in a three-dimensional network.

**Table 1**

Selected geometric parameters ( $\text{\AA}$ ).

O1–C7	1.216 (2)	N1–C6	1.443 (2)
O2–C14	1.208 (2)	C7–C8	1.485 (2)
N1–C7	1.411 (2)	C14–C15	1.485 (2)
N1–C14	1.426 (2)		

**Table 2**

Intermolecular contacts ( $\text{\AA}$ ,  $^\circ$ ).

$D\text{--}H\cdots A$	$D\text{--}H$	$H\cdots A$	$D\cdots A$	$D\text{--}H\cdots A$
C9–H9...O1 <sup>i</sup>	1.00 (2)	2.47 (2)	3.130 (2)	123 (1)
C13–H13...O3 <sup>ii</sup>	1.00 (2)	2.49 (2)	3.380 (2)	148 (1)
C22–H22B...O1 <sup>iii</sup>	0.96	2.46	3.335 (2)	151
C1–H1...Cg(B <sup>i</sup> )	0.97 (2)	2.624 (2)	3.520 (2)	153 (1)
C10–H10...Cg(C <sup>i</sup> )	0.98 (2)	2.692 (2)	3.344 (2)	124 (1)

Symmetry codes: (i)  $\frac{1}{2} - x, y - \frac{1}{2}, \frac{1}{2} - z$ ; (ii)  $x, 1 + y, z$ ; (iii)  $2 - x, 1 - y, -z$ .

The H atoms were located from a difference Fourier map and were refined isotropically, with the exception of the H atoms attached to C22; these were geometrically placed and were treated as riding atoms, with  $C\text{--}H = 0.96 \text{ \AA}$  and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ .

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*, *PARST* (Nardelli, 1995) and *PLATON* (Spek, 1990).

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